# Silent Starter

https://www.ocr.org.uk/l mages/171726specification-accrediteda-level-gce-physics-ah556.pdf



#### 2 The specification overview

#### 2a. Overview of A Level in Physics A (H556)

Learners must complete all components (01, 02, 03 and 04) to be awarded the OCR A Level in Physics A.

<b>Content Overview</b>	Assessment Overview		
<ul> <li>Content is split into six teaching modules:</li> <li>Module 1 – Development of practical skills in physics</li> </ul>	Modelling physics (01) 100 marks 2 hours 15 minutes written paper	<b>37%</b> of total A level	
<ul> <li>Module 2 – Foundations of physics</li> <li>Module 3 – Forces and motion</li> <li>Module 4 – Electrons, waves and photons</li> </ul>	Exploring physics (02) 100 marks 2 hours 15 minutes written paper	<b>37%</b> of total A level	
<ul> <li>Module 5 – Newtonian world and astrophysics</li> <li>Module 6 – Particles and medical physics</li> <li>Component 01 assesses content from modules 1, 2, 3 and 5.</li> </ul>	Unified physics (03) 70 marks 1 hour 30 minutes written paper	<b>26%</b> of total A level	
Component 02 assesses content from modules 1, 2, 4 and 6. Component 03 assesses content from all modules (1 to 6).	Practical Endorsement in physics (04) (non exam assessment)	Reported separately (see Section 5g)	

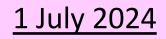


## Silent Starter





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Learning objectives Describe the standard model Explain the structure of particles

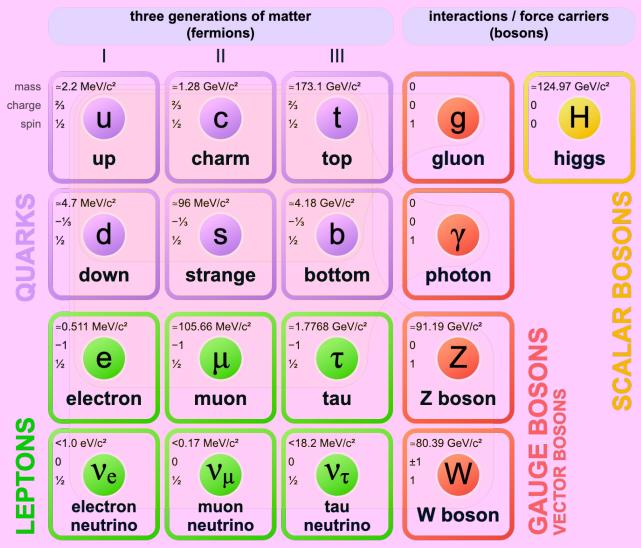
Key words: Quark Lepton Hadron Baryon Proton Neutron Electron



Any particle that contains quarks is a hadron.

A lepton is a fundamental particle



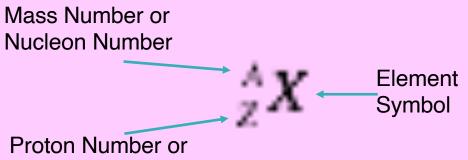




# The Nucleus

The nucleus consists of protons and neutrons, which are collectively known as nucleons.

The following notation represents the nucleus of the atom:



Electrons Electrons Nucleons within positive nucleus

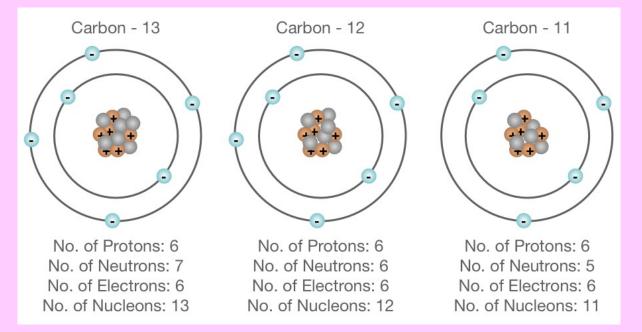
Proton Number or Atomic Number





# Isotopes

An isotope is any of two or more forms of a chemical element. They have the same number of protons in the nucleus, but have different numbers of neutrons.





# Constituents of the Atom

### Proton

	Relative	SI Unit
Mass	1	1.673 x 10 <sup>-27</sup> kg
Charge	1	1.6 x 10 <sup>-19</sup> C

### Neutron

	Relative	SI Unit
Mass	1	1.675 x 10 <sup>-27</sup> kg
Charge	0	0

### Electron

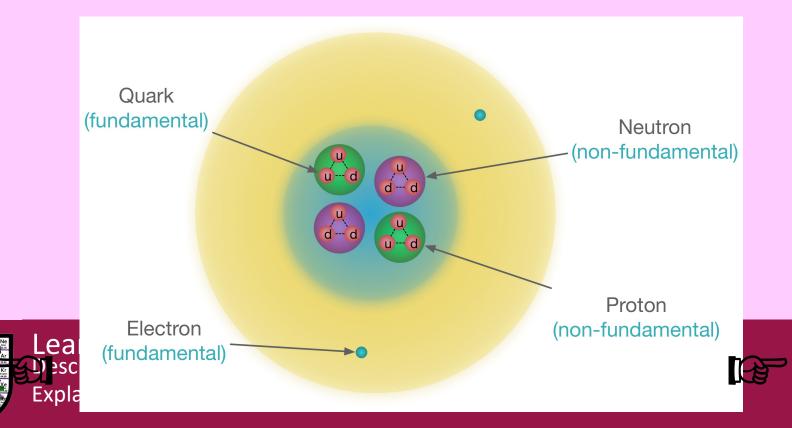
	Relative	SI Unit	
Mass	0.0005	9.11 x 10 <sup>-31</sup> kg	
Charge	-1	-1.6 x 10 <sup>-19</sup> C	





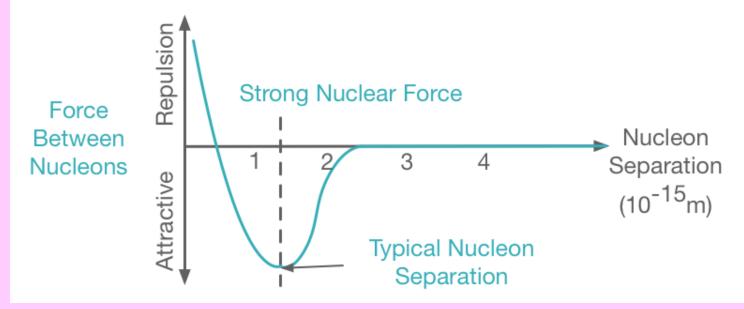
# Fundamental Particles

Fundamental (elementary) particles are those which are not composed of other particles. The atom consists of fundamental and non-fundamental particles



# **Strong Force**

### The nucleus is held together by the strong force.



- Highly attractive between two protons within 1-2 femtometres
- Repulsive between two protons at smaller separations (cannot overlap)

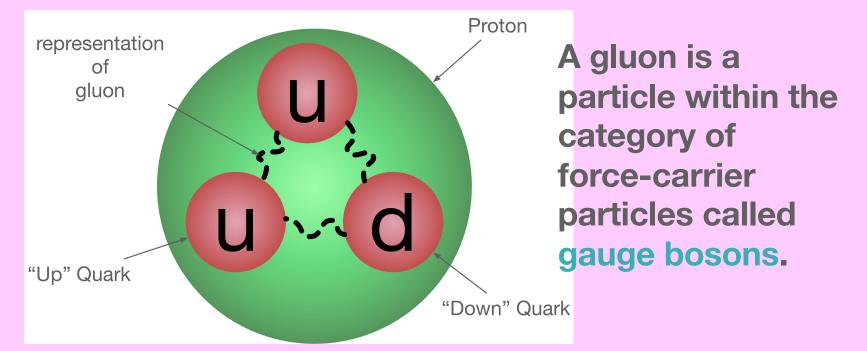






# **Explanation** Exchange Particles

It is thought that exchange particles, called gluons, are responsible for the Strong Force. They act between the quarks in a neutron or proton.





# Explanation Four Fundamental Forces

There are four fundamental forces that act within a nucleus. Each is thought to have their own set of exchange particles, which "carry" the force:

Fundamental Force:	Acts On:	Exchange particles (gauge boson):
Strong	Quarks	Gluons
Gravitational		Gravitons (not yet observed)
Electromagnetic	All particles with charge	Photons
Weak	Leptons	W and Z Bosons

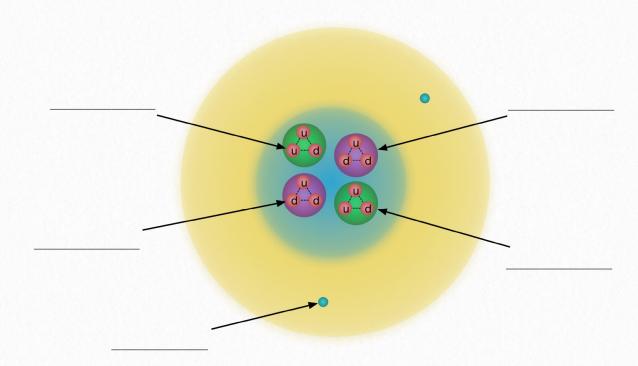




Complete the Questions

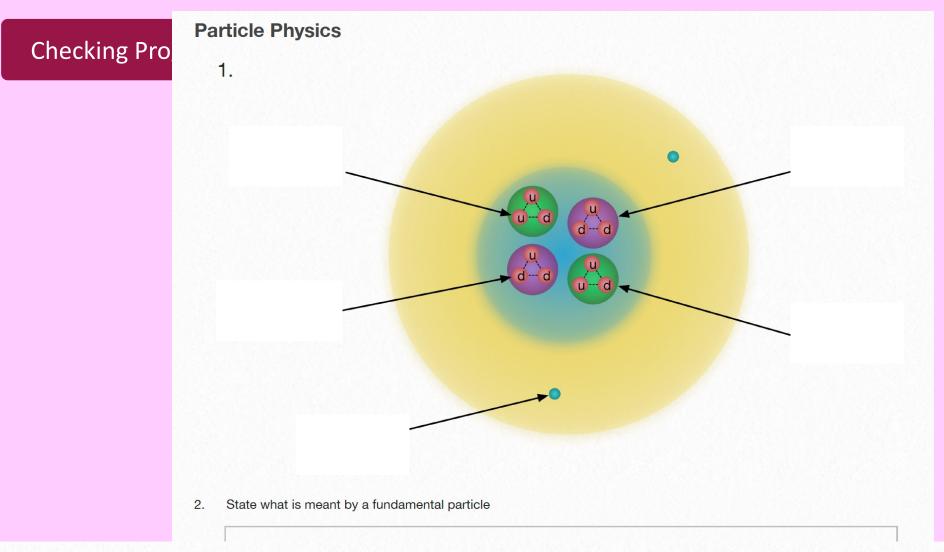
#### **Particle Physics**

1. Label the diagram of the particles within the atom below:



2. State what is meant by a fundamental particle



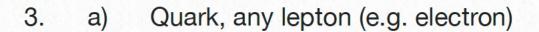


Fundamental (elementary) particles are those which are not composed of other particles.

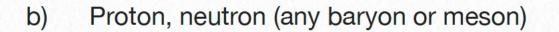


#### Checking Progress

- 3. The particles within an atom can be divided into fundamental particles and non-fundamental particles.
  - a) Give two examples fundamental particles:



b) Give two examples of non-fundamental particles:

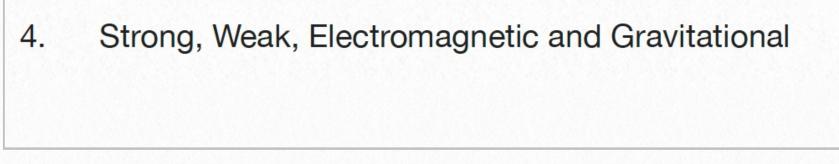




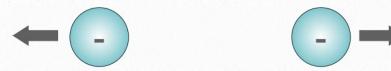
# **Questions Continued**

#### **Particle Physics**

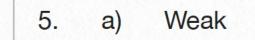
4. Name the four fundamental forces.



5. Two electrons approach each other but do not collide. They exert a force on each other and move apart.



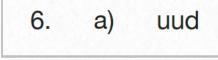
a) Which of the four fundamental forces is involved in this process.



b) Name the exchange particle that plays a role in this interaction.

b) Photon (Gauge Boson)

- 6. State the quark composition of:
  - a) The proton



b) The neutron



7. A  $\pi^o$  particle is classed as a meson. It has a charge of 0 and a baryon number of 0. Using the quark table below, which of the following combinations could correspond to a  $\pi^o$  meson.

A.	sū			Quark	Charge
B.	udd	7.	С	u	+2/3
C.	$d\bar{d}$			d	-1/3
D.	иđ			S	-1/3



# End & send

# **Expectations**

- 1. Put the borrowed <u>equipment</u> back.
- 2. Bin all <u>rubbish</u>.
- 3. Put your belongings in your <u>bag</u>.
- 4. <u>Double check 1 to 3 has been done.</u>
- 5. <u>Stand</u> and <u>stay</u> behind your <u>chair silently</u>.



